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LEWIS ACID INITIATED OR HIGH PRESSURE PROMOTED REACTIONS OF ISOPRENE WITH PHENYISULPHINYL CHLORIDE

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> <u>Summary</u>. Lewis acid initiated reaction of isoprene with PhSOC1 proceeds in a ene fashion with the formation of 2-phenylsulphinylmethyl-1,3-butadiene. High pressure promoted addition produces Z-1-phenylsulphinyl-4-chloroadduct presumably <u>via</u> [4+2] cycloaddition.

According to existing scarce reports,¹ Ad_E -reaction of ArSOC1 with alkenes proceeds only in the presence of Lewis acids and yields respective 1, 2-arylsulphinyl chloroadducts. No data are available on the reactivity pattern of 1,3-dienes in this reaction. In the search of novel methods for the regiospecific isoprene (<u>1</u>) functionalization we have studied the interaction of <u>1</u> with PhSOC1 under various conditions.

The addition of PhSOCl to <u>1</u> was found to proceed smoothly in the presence of Lewis acids (e.g. $2nCl_2$, $SnCl_4$, $TiCl_4$, $BF_3 \cdot Et_20$) in CH_2Cl_2 or $MeNO_2$ at -20 + 0°C producing the mixture of diene <u>2</u> and E,Z-chloroolefins <u>3,4</u> (Scheme) their separation being easily achieved by flash chromatography $(SiO_2, gradient elution from hexane to ether). Under optimal conditions <math>(-20°C, MeNO_2, 1 : PhSOCl:2nCL_2 = 3:1:1)$ <u>2</u> is formed as a major product $(\underline{2:2:4} \approx 15:5:1, \text{ total yield 80\%})$. Similar reaction performed with AgBF₄ instead of $2nCl_2$ gave <u>2</u> as a single product² with <u>ca</u>. 75% isolated yield (b.p. 100-103°C/0.05 torr; n_D^{20} 1.5873; PMR, 250 MHz, $\delta_{TMS}^{CDCl_3}$: 3.69 ppm,q, AB, CH_2S). Sulphoxide <u>2</u> was transformed with MCPBA into the known³ sulphone <u>10</u> (m.p.65-65.5°C, from ether-hexane; δ 4.000 s, CH_2S).^{2,4} E-Configuration of the main isomer of1,4-adduct <u>3</u> (m.p. 65-66°C from ether-hexane; δ 3.50 bs, CH_2S , 4.11 d, J=7.8 Hz, CH_2Cl)² was checked by its MCPBA oxidation to the

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Scheme



known⁵ E-sulphone 2. The structure of the minor isomer <u>4</u> will be discussed below.

To account for the exclusive formation of 2 in AgBF₄ initiated reaction it can be suggested that the addition of incipient phenylsulphinyl cation to 1,2-double bond of <u>1</u> proceeds in an ene-like fashion <u>via</u> six-membered transition state <u>5</u>. Similar mechanism is evidently operating in $2nCl_2$ induced reaction. Adducts <u>3</u> and <u>4</u> formed in the latter case probably arose due to secondary process involving the initial formation of 1,2-adduct <u>7</u> followed by its non-stereospecific isomerization into the mixture of 1,4-adducts in the presence of Lewis acid (<u>c.f.</u>⁶ similar 1,2--- 1,4 isomerisation of ArSC1 -isoprene adducts <u>6</u>).

An ever increasing number of heterodienic reactions involving various O_1N , and S containing dienophiles⁷ and ene-like reactivity of PhSOCl observed made it very tempting to look for the conditions of its [4+2] reaction with <u>1</u>. It turned out however that in the abscence of Lewis acids the interaction of <u>1</u> and PhSOCl in various solvents (CH₂Cl₂, Et₂O, THF, glyme, etc.)

proceeds very slowly at ambient temperature ($\leq 25\%$ of convergion after one week) giving rise rather complicate product mixture. Our attempts to carry out the reaction under more drastic conditions (up to 100° C) failed due to thermolability of PhSOC1.

The benefitial effects of high pressure on the rate and selectivity of [4+2] cycloadditions are well-known.⁸ These data encouraged us to test the effect of high pressure on the reaction under study. The result observed was rather unexpected and truly rewarding.

We have found that the application of high pressure (3-5 kbar) to the solution of 2:1 mixture of <u>1</u> and PhSOCl in CHCl₃ resulted in <u>ca</u>. 75% yield formation of Z-chlorosulphoxide <u>4</u> uncontaminated by the admixture of <u>3</u> or <u>2</u> (250 MHz PMR data). In all runs PMR data revealed the presence of 7-15% of unknown impurity; fortunately, the latter is unstable and is easily removed in the course of further chemical transformation of <u>4</u> (vide infra).

The structure of $\underline{4}$ was supported by its PMR spectrum (δ 3.59, AB-quartet, CH₂S and 3.87 ppm, AB-part of ABX system, CH₂Cl). The comparison of CMR spectrum (62.08 MHz, CDCl₃) of $\underline{4}$ with that of E-isomer 3 revealed the substantial upfield shift of CH₂S signals in the former (δ 60.6 for $\underline{4}$ and 67.7 ppm for 3). Both $\Delta \delta$ sign and its magnitude are in a well accord with the stereochemistry ascribed to these compounds.⁹ MCPBA oxidation of $\underline{4}$ yields the known³ Z-sulphone 8, m.p. 88-89°C (from ether-hexane).^{2,4} The same shift, $\Delta \delta \simeq 7$ ppm, was observed for CH₂S signals within the pair 9 and 8.

The mechanism of high pressure promoted formation of $\underline{4}$ is at present unclear. We believe however that the observed Z-stereospecificity of its formation is certainly indicative of some kind of cycloaddition process involved. One may suggest that the addition of PhSOC1 to $\underline{1}$ proceeds in a [4+2] fashion <u>via</u> six-membered $\underline{\circ}$ -sulphurane intermediate $\underline{11}$. Intrinsic instability of the latter precludes its isolation; the proposed isomerization $\underline{11}$ - $\underline{4}$ has some analogy with the chemical transformations observed for the stabilized $\underline{\circ}$ -sulphuranes.¹⁰

Easily accessible chlorosulphoxide $\underline{4}$ represents a novel Z-isoprenoid synthon which could be converted to a series of related functionalized derivatives. For example, acetolysis of $\underline{4}$ yields (~75%) oily sulphoxide $\underline{12}$ (δ 3.65, AB-quartet, CH₂S, and 4.32 ppm, AB-part of ABX system, CH₂O).² Reduction of <u>12</u> with IAH at -20°C proceeds selectively to give 70% yield of hydroxy sulphoxide <u>13</u> as an oil; its spectral data matched those reported earlier¹¹ for this compound. Similar reduction at 25°C produces (~ 75%) oily hydroxy sulphide <u>14</u> (b.p. 96-98°C/0.07 torr; n_D²⁰1.5840; δ 1.88 bs, CH₃, 3.50 s, CH₂S, 3.78 bd, J=7.0 Hz, CH₂O, 5.48 bt, HC=C).²

Thus, the reaction of <u>1</u> with PhSOC1 could serve as the selective preparative method for the synthesis of either diene <u>2</u> or Z-adduct <u>4</u>. Both products and derived diene <u>10</u> and Z-1,4 -bifunctional compounds <u>12-14</u> are well suited for their further use as the isoprenoid synthons. Preliminary studies also indicated that diene <u>10</u> is able to react with typical dienophyles (e.g. maleimides) and hence could be used for the synthesis of cyclic isoprenoids <u>via</u> [4+2] reaction with suitable substrates.

Our further work is aimed at the evaluation of applicability of the disclosed 1,2-ene and Z-1,4-stereoselective additions to other 1,3-dienic systems.

References and Notes

- 1. cf. B.B.Snider, J.Org. Chem. 46, 3155 (1981) and references therein.
- 2. Satisfactory analytical data were obtained for this compound.
- 3. P.J.R.Nederlof, M.J.Moolenaar, E.R. de Waard, H.O.Huisman, Tetrahedron, <u>34</u>, 447 (1978).
- 4. In ref.³ m.p. 52-53°C is given for 10 and m.p. 78-80°C for 8.
- 5. W.E.Truce, C.T.Goralski, L.W.Christensen, R.H.Bavry, J.Org.Chem. <u>35</u>, 4217 (1970).
- 6. G.Kresze, W.Kosbahn, Lieb.Ann.Chem. 731, 67 (1970).
- 7. S.M.Weinreb, R.R.Staib, Tetrahedron, <u>38</u>, 3087 (1982).
- 8. V.M.Zhulin, S.I.Volchek, Khim.Geterotsikl.Soedin., 1979, 1443.
- 9. F.Bohlmann, R.Zeisberg, E.Klein, Organ.Magn.Resonance, 7, 426 (1975).
- 10. J.C.Martin, E.F.Perozzi, Science, <u>191</u>, 154 (1976).
- 11. P.J.R.Nederlof, M.J.Moolenaar, E.R. de Waard, H.O.Huisman, Tetrahedron Lett., <u>1976</u>, 3175.

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